



Ag/Ce_{1-x}Mn_xO₂ Catalysts for Soot Oxidation: Role of Ce/Mn Molar Ratio

Ekaterina S. L'vova,¹ Tamara S. Kharlamova,¹ Maria V. Grabchenko,¹ Olga V. Vodyankina,^{*1} Eleonora La Greca,² and Leonarda F. Liotta^{*2}

¹ Tomsk State University, 36 Lenin Ave., Tomsk, 634050, Russian Federation

² CNR, ISMN, Via Ugo La Malfa 153, 90146-Palermo, Italy

* leonardafrancesca.liotta@cnr.it

Significance and Relevance

This study highlights the development of Ag/CeO₂, Ag/MnO_x, and Ag/CeMnO_x catalysts prepared via the citrate sol-gel method, focusing on their efficacy in soot oxidation. The novel nanodomain structure in CeMnO_x, influenced by the Ce/Mn ratio, enhances catalytic activity and stability, particularly at a 1:1 ratio, which prevents CeO₂ sintering. The catalysts demonstrate selective soot oxidation to CO₂ at relatively low temperatures (<500°C for powders and <600°C for monoliths), showcasing their practical potential. Additionally, the findings reveal NO inhibitory effects on oxidation, mitigated by CO in NO+O₂ mixtures, offering valuable insights for emission control technologies.

Preferred and 2nd choice for the topic: Automotive and stationary emission control; Air cleaning and combustion.

Preferred presentation: Oral preferred or Short Oral

Introduction and Motivations

Diesel engines, known for their strong power performance and high fuel efficiency, are considered a promising power source for light vehicles, which have predominantly relied on gasoline engines. However, a widespread adoption of diesel engines is hindered by significant emission issues, particularly the release of particulate matter (PM) mainly composed of soot. The use of catalytic filters, specifically diesel particulate filters (DPFs), which trap and oxidize soot effectively reducing its harmful impact by significantly lowering the soot oxidation temperature. CeO₂ is highly regarded for its exceptional catalytic activity and redox behavior.¹ Similarly, MnO_x is widely utilized in oxidation and reduction reactions, thanks to its multivalent oxidation states (mainly Mn²⁺, Mn³⁺, and Mn⁴⁺) and the high mobility of lattice oxygen.² When combined, MnO_x-CeO₂ catalysts demonstrate superior catalytic activity significantly lowering the ignition temperature for soot oxidation due to the synergistic effect between the two metal oxides.³ Furthermore, the addition of silver to the CeO₂ and MnO_x catalyst significantly boosts its reactivity for soot oxidation.⁴ The aim of this study was to evaluate the performance of the Ag/CeMnO_x catalysts in soot combustion under tight and loose contact conditions flowing 10 vol.% O₂/He as well as in the presence of NO and NO-CO.

Materials and Methods

Oxide supports were prepared by the citrate sol-gel method, using Ce(NO₃)₃ · 6H₂O and Mn(NO₃)₂ · 6H₂O as precursors in Ce/Mn molar ratios equal to 1/3, 1/1 or 3/1, after drying were calcined at 500°C. Ag catalysts (1 wt.%) were prepared by wet impregnation using Ag(NH₃)₂ and were calcined at 500 °C. The resulting samples were labeled as Ag/Ce, Ag/Mn and Ag/1Ce3Mn, Ag/1Ce1Mn and Ag/3Ce1Mn. Portions of the catalysts were treated in oven at 650°C for 12 h in order to simulate thermal aging. Characterization by XRD, N₂ physisorption, XRF, Raman, SEM and TEM techniques was performed. Catalytic tests were carried out under tight and loose contact conditions in temperature-programmed

reaction (TPR) mode, under a flow of 10 vol.% O₂/He, 10 vol.% O₂, 0.1 vol.% NO/He, or 10 vol.% O₂, 0.1 vol.% NO, 1.25 vol.% CO/He.

Results and Discussion

Fig. 1a shows that in the presence of the supported Ag catalysts the temperature of soot oxidation significantly decreased with respect to the not catalyzed process. Moreover, in the presence of catalysts soot was fully oxidized to CO₂, while without any catalyst CO and CO₂ are formed. The Ag/Mn and Ag/1Ce3Mn samples demonstrated the lowest activity (~50% of soot conversion was achieved at around 500°C). A further increase in the Ce content in the oxide matrix resulted in the improvement of the catalyst activity, the Ag/Ce sample showing the highest activity (T_{max}=448°C). Thermal aging at 650°C for 12 h lead to a significant deactivation of the Ag/Ce (aged) (T_{max}=529°C) due to a strong decrease of the specific surface area and an increase in the crystallites size. The T_{max} of soot oxidation over Ag/3Ce1Mn (aged) increased to 500°C (ΔT_{max}=37° C), while the Ag/1Ce1Mn (aged) appeared the most stable with the best oxidation performance (T_{max}=465°C, ΔT_{max}=7° C), (Fig. 1b).

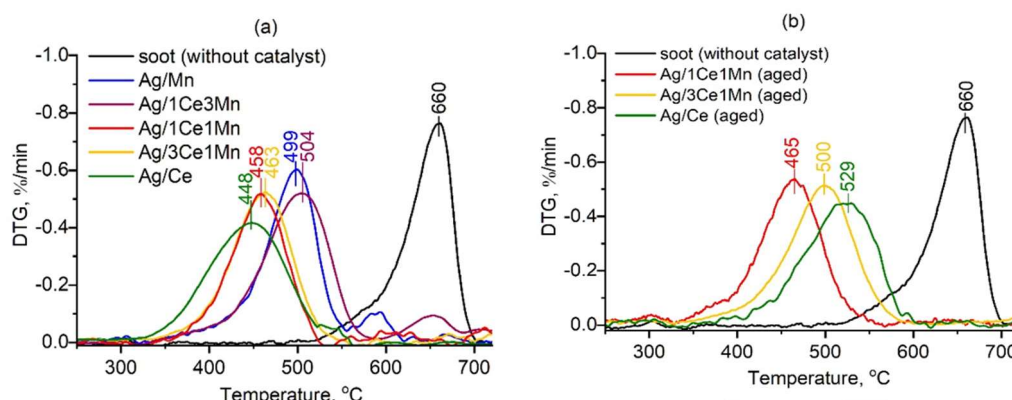


Figure 1. Soot oxidation activity versus temperature for fresh Ag supported catalysts (a) and those aged at 650 °C for 12 h (b).

References

1. A. Setiabudi, J. Chen, G. Mul, M. Makkee, J.A. Moulijn, *Applied Catalysis B Environmental* **2004**, 51(1), 9-19
2. R. Mane, H. Kim, K. Han, K.-J. Kim, S. S. Lee, H.-S. Roh, C. Lee, Y. Jeon, *Fuel* **2023**, 346, 128287
3. X. Lin, S. Li, H. He, Z. Wu, J. Wu, L. Chen, D. Ye, M. Fu, *Applied Catalysis B Environmental* **2018**, 223, 91-102.
4. M.V. Grabchenko, G.V. Mamontov, V.I. Zaikovskii, V. La Parola, L.F. Liotta, O.V. Vodyankina, *Applied Catalysis B Environmental* **2020**, 260, 118148

Acknowledgements

This work was supported by the Russian Science Foundation, Grant No. 19-73-30026. The authors acknowledge A.G. Golubovskaya (Tomsk State University) for SEM study.