

Recyclable Ni@Al₂O₃ catalyst for Dry Reforming of Methane

Rachele BRAIDO^{*,1}, Giulia DA PIAN,¹ Irene MARTIN,² Giuseppe CRUCIANI,³ and Michela SIGNORETTO¹

¹ Department of Molecular Sciences and Nanosystems, Ca' Foscari University of Venice, and INSTM RUVe Via Torino 155, 30172-Venezia, Italia

² Functional Nanosystems, Italian Institute of Technology (IIT), Via Morego 30, 16163 Genoa Italy, and Department of Applied Science and Technology, Politecnico di Torino, Corso Duca degli Abruzzi 34, 10129 Turin Italy

³ Department of Physics and Earth Science, University of Ferrara, Via Saragat 1, 44122-Ferrara, Italia * <u>rachele.braido@unive.it</u>

Significance and Relevance

A Ni@Al₂O₃ catalyst with a core-shell like structure was tested for Methane Dry Reforming (DRM). Studies suggest that protecting the metal's active site can prevent deactivation caused by sintering and coke deposition. In this study stability tests and several regeneration cycles were performed, demonstrating that the different catalytic structure obtained is suitable for DRM purposes.

Preferred and 2^{*nd*} *choice for the topic:* H₂ storage and transportation, green H₂ production, hydrogen vectors, CO₂ utilization and recycling

Preferred presentation: Poster

Introduction and Motivations

Dry reforming of Methane is a catalytic reaction that converts two of the most impactful greenhouse gases - CH_4 and CO_2 - into syngas¹. The reaction has not been applied yet at industrial level due to two major deactivation challenges: sintering of the metal active phase and coke deposition. The first one is related to the high temperature required by the reaction's endothermic nature. The second one is related to secondary side reactions². To address the deactivation problems, the metal active site (Ni) can be surrounded by a metal oxide shell (AI_2O_3), which protects the Ni and provides a confinement effect.

Materials and Methods

The obtained catalyst was tested for the DRM reaction at 650 °C in a fixed bed reactor with a GHSV of 12000 h^{-1} , using a CH₄:CO₂:He ratio of 1:1:18, under a total flow of 200 mL/min. Additionally, recycling tests were conducted.

The fresh and spent samples were characterized using various techniques such as XRD, SEM, TEM, TPR, TPO, and N_2 physisorption.

Results and Discussion

The activity test demonstrated that the catalyst has an active and stable catalytic behaviour achieving a good H_2/CO ratio. In addition, the recycling test showed that the catalyst effectively converts the reactant gases into syngas, being stable for up to five redox cycles.

References

 M. Pizzolato, G. Da Pian, E. Ghedini, A. Di Michele, F. Menegazzo, G. Cruciani, and M. Signoretto, *Reactions*, 2022, *3*, 634 – 647.
M. Pizzolato, G. Da Pian, E. Ghedini, A. Di Michele, F. Menegazzo, G. Cruciani, and M. Signoretto, *Catalisys Today*, 2023, *418*, 114041.